

Uncertainty in the Volume of a Chocolate-Chip cookie.

$$D = 8.50 \pm 0.02 \text{ cm} \quad h = 8.5 \times 10^{-2} \pm 0.005 \text{ cm.}$$

$$V = \frac{\pi D^2}{4} h = 4.823 \text{ cm}^3$$

$$dV = \frac{\pi}{4} (2D dD h + D^2 dh) = \frac{\pi(D^2 h)}{4} \left(2 \frac{dD}{D} + \frac{dh}{h} \right)$$

$$dV = V \left(2 \frac{dD}{D} + \frac{dh}{h} \right)$$

① Using this approach $\Rightarrow dV_1 = 0.306 \text{ cm}^3$

\Rightarrow ② Using the statistical approach $dV_2 = 0.285 \text{ cm}^3$

$$dV_2 = V \sqrt{\left(2 \frac{dD}{D} \right)^2 + \left(\frac{dh}{h} \right)^2}$$

$$\frac{dV_2}{V} = 5.90\%$$

Why is the 2nd technique the correct one?

Using the 1st technique (i.e., the bare-bones calculus approach)

Imagine $V = \frac{x}{t} \quad dv = \frac{1}{t} dx + \left(\frac{-x}{t^2} \right) dt$

uncertainty in the velocity $\Rightarrow dv = \frac{x}{t} \left(\frac{dx}{x} - \frac{dt}{t} \right)$ using the 1st approach.

could = 0 ?? How can that be?? I can't!!